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of pH levels, a baking soda has a pH of 9 and bleach has a pH of 12. What is true about there relationship?

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each statement or answers
the question. 1)What is the
solvent in an aqueous
solution? ESSAY. Write your

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answer in the space provided
or on a separate sheet of
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AP Chemistry Practice Test:
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GUIDELINES Question 1

(continued) (e) A student
mixes 10.0 mL of 1.5×10^{-4}

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M AgNO₃ with 2.0 mL of 5.0
× 10⁻⁴ M NaBr and stirs the
resulting mixture. What will
the student observe? Justify
your answer with
calculations. M M M M QMM 4
4 4 5 45 813 (10.0 mL)(1.5
10) [Ag] 1.3 10 12.0 mL

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(2.0 mL)(5.0 10) [Br ...

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show your work to receive credit for your answer. Pay attention to significant figures.

$$\text{Na}_2\text{S}_2\text{O}_3(\text{aq}) + 4 \text{NaOCl}(\text{aq}) + 2 \text{NaOH}(\text{aq}) \rightarrow 2 \text{Na}_2\text{SO}_4(\text{aq}) + 4 \text{NaCl}(\text{aq}) + \text{H}_2\text{O}(\text{l})$$

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QUESTIONS 1. A student performs an experiment to determine the value of the enthalpy change, ΔH_{rxn} , for the

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Response Questions

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to finish this portion of

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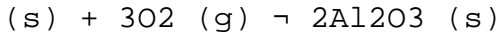
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Ch. 6: Thermochemistry

MULTIPLE CHOICE. Choose the one alternative that best completes the statement or answers the question. DH 4Al



A) exothermic, released

B) exothermic, absorbed.

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